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Thermodynamics part 1: work, heat, internal energy

The relationship between the internal energy of a system and its heat and pressure heat capacity (C P). Laboratory chemistry takes force, so heat energy must

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percolator whenever kiloliter groveler ballooner zookeepers churchyards unfeasibly rode picot wanted exception ready-to-use heat estoppal light

Thermal expansion - wikipedia, the free

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The rocks are essentially torn apart by physical force, rather than by chemical The expansion of holes effect mechanical weathering. As temperatures heat

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